# metal-organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

## Hong-Jun Yang,<sup>a</sup> Lai-Jin Tian,<sup>a</sup>\* Chang-Fa Zhang<sup>b</sup> and Xiao-Feng Zheng<sup>a</sup>

<sup>a</sup>Department of Chemistry, Qufu Normal University, Qufu 273165, People's Republic of China, and <sup>b</sup>Shandong Water Polytechnic, Rizhao 276826, People's Republic of China

Correspondence e-mail: laijintian@163.com

#### **Key indicators**

Single-crystal X-ray study T = 295 KMean  $\sigma(\text{C-C}) = 0.006 \text{ Å}$  R factor = 0.031 wR factor = 0.078 Data-to-parameter ratio = 17.1

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

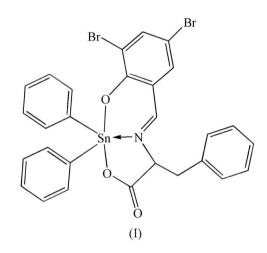
# [*N*-(3,5-Dibromo-2-oxidobenzylidene)phenylalaninato- $\kappa^3 O, N, O'$ ]diphenyltin(IV)

The Sn atom of the title compound,  $[Sn(C_6H_5)_2(C_{16}H_{11}-Br_2NO_3)]$ , adopts a distorted  $SnNC_2O_2$  trigonal-bipyramidal geometry and forms five- and six-membered chelate rings with the tridentate ligand.

Received 28 March 2007 Accepted 28 March 2007

## Comment

The structural chemistry of diorganotin complexes with Schiff bases derived from  $\alpha$ -amino acids continues to receive attention owing to their biological properties, especially their antitumour activities (Beltran *et al.*, 2003; Dakternieks *et al.*, 1998; Tian *et al.*, 2005, Tian *et al.*, 2006, 2007; Yin *et al.*, 2004). As a continuation of these studies, the synthesis and structure of the title compound, (I), are now described.



The coordination geometry about the tin atom in (I) is distorted trigonal bipyramidal, with two phenyl group C atoms (C17 and C23) and the imino N1 atom occupying the equatorial positions. The axial positions are occupied by the unidentate carboxylate atom O1 and the phenoxide atom O2 (Fig. 1). The Sn-O2 bond length is significantly longer than Sn-O1, and the O1-Sn-O2 angle is 157.50 (8)° (Table 1). The monodentate coordination mode of the carboxylate group is reflected in the disparate C9-O2 and C9-O3 bond lengths of 1.292 (4) and 1.212 (4) Å, respectively. Otherwise, the geometrical parameters for (I) are comparable to those observed in the diphenyltin complexes referenced above.

## **Experimental**

The title compound was synthesized by the reaction of diphenyltin(IV) dichloride (0.69 g, 2 mmol) with potassium N-(3,5-dibromosalicylidene)phenylalaninate (0.93 g, 2 mmol), derived from potassium hydroxide, L-phenylalanine and 3,5-dibromo-

\_\_\_\_\_

© 2007 International Union of Crystallography

All rights reserved

m1272

salicylaldehyde **[amounts of the three reagents?]** in the presence of  $Et_3N$  (0.20 g, 2 mmol) in methanol (60 ml). The reaction mixture was refluxed for 3 h, and then cooled to room temperature and filtered. The yellow solid obtained by removal of solvent under reduced pressure was recrystallized from ethanol and yellow plates of (I) were obtained by slow evaporation from chloroform–hexane (1:1  $\nu/\nu$ ) at 298 K (yield 67%; m.p. 393–394 K). Analysis found: C 48.31, H 3.00, N 2.03%; calculated for  $C_{28}H_{21}Br_2NO_3Sn:$  C 48.18, H 3.03, N 2.01%.

V = 2609.4 (3) Å<sup>3</sup>

Mo  $K\alpha$  radiation  $\mu = 4.07 \text{ mm}^{-1}$ T = 295 (2) K

 $R_{\rm int} = 0.033$ 

 $0.35 \times 0.25 \times 0.08 \ \text{mm}$ 

20999 measured reflections

5406 independent reflections

4566 reflections with  $I > 2\sigma(I)$ 

Z = 4

## Crystal data

| $[Sn(C_6H_5)_2(C_{16}H_{11}Br_2NO_3)]$ |
|--|
| $M_r = 697.97$                         |
| Monoclinic, $P2_1/c$                   |
| a = 11.2880 (7)  Å                     |
| b = 23.3977 (15) Å                     |
| c = 10.7391 (7)  Å                     |
| $\beta = 113.074 \ (1)^{\circ}$        |

### Data collection

Bruker SMART APEX CCD diffractometer Absorption correction: multi-scan (SADABS; Bruker, 2002)  $T_{min} = 0.330, T_{max} = 0.737$ 

### Refinement

| $R[F^2 > 2\sigma(F^2)] = 0.031$ | 316 parameters   |
|---------------------------------|--|
| $wR(F^2) = 0.078$               | H-atom parameters constrained                              |
| S = 1.04                        | $\Delta \rho_{\rm max} = 0.80 \ {\rm e} \ {\rm \AA}^{-3}$  |
| 5406 reflections                | $\Delta \rho_{\rm min} = -0.44 \text{ e } \text{\AA}^{-3}$ |

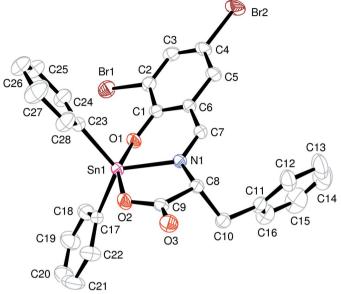
### Table 1

Selected geometric parameters (Å, °).

| Sn1-O1      | 2.095 (2)   | Sn1-O2     | 2.114 (2)   |
|-------------|-------------|------------|-------------|
| Sn1-C17     | 2.107 (3)   | Sn1-N1     | 2.168 (2)   |
| Sn1-C23     | 2.110 (3)   |            |             |
| O1-Sn1-C17  | 93.28 (10)  | C23-Sn1-O2 | 96.06 (11)  |
| O1-Sn1-C23  | 97.24 (10)  | O1-Sn1-N1  | 82.35 (8)   |
| C17-Sn1-C23 | 121.98 (11) | C17-Sn1-N1 | 122.87 (10) |
| O1-Sn1-O2   | 157.50 (8)  | C23-Sn1-N1 | 115.06 (10) |
| C17-Sn1-O2  | 94.95 (11)  | O2-Sn1-N1  | 75.59 (8)   |

H atoms were positioned geometrically (C-H = 0.93–0.98 Å) and refined as riding, with  $U_{iso}(H) = 1.2U_{eq}(C)$ .

Data collection: *SMART* (Bruker, 2002); cell refinement: *SAINT* (Bruker, 2002); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics:



#### Figure 1

The molecular structure of (I), with displacement ellipsoids drawn at the 30% probability level and H atoms omitted for clarity.

*ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *SHELXL97*.

The authors thank the Science Foundation of Shandong Province and Qufu Normal University for supporting this work.

## References

- Beltran, H. I., Zamudio-Rivera, L. S., Mancilla, T., Santillan, R. & Farfan, N. (2003). Chem. Eur. J. 9, 2291–2306.
- Bruker (2002). SADABS, SAINT and SMART. Bruker AXS Inc., Madison, Wisconsin, USA.
- Dakternieks, D., Basu Baul, T. S., Dutta, S. & Tiekink, E. R. T. (1998). Organometallics, 17, 3058–3062.
- Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Tian, L., Qian, B., Sun, Y., Zheng, X., Yang, M., Li, H. & Liu, X. (2005). Appl. Organomet. Chem. 19, 980–987.
- Tian, L., Shang, Z., Zheng, X., Sun, Y., You, Y., Qian, B. & Liu, X. (2006). Appl. Organomet. Chem. 19, 74–80.
- Tian, L., Sun, Y., Zheng, X., Liu, X., You, Y., Liu, X. & Qian, B. (2007). Chin. J. Chem. 25, 312–318.
- Yin, H., Wang, Q. & Xue, S. (2004). J. Organomet. Chem. 689, 2480-2485.